Name: _	
Date:	

#### **Metric Measurement Lab**

## Objectives:

- You will learn to make measurements using the metric system.
- These measurements will encompass mastering the metric ruler, gram scale, and the graduated cylinder, and thermometer.
- You will demonstrate your ability to convert the original measurements to lower and higher values by moving the decimal point the correct number of places in the proper direction.

#### Materials:

- meter stick
- metric ruler
- graduated cylinders, beaker
- coins
- book
- electronic balance/scale
- thermometer

## **Procedures and Data:**

#### 1. Linear Measure

Use the metric rule <u>OR</u> meter stick to measure the items listed below. Place your measurements in the spaces below. Above each column write the name of the unit that is abbreviated below it. Circle the unit you used to measure with for each item. You will need to convert for the other units.



Millimeters

Diameter of Penny	m.	cm.	mm.	km.
Height of lab counter	m.	cm.	mm.	km.
Width of the Textbook	m.	cm.	mm.	km.
Length of the Room	m.	cm	mm.	km.

1a. Did you use the same unit to measure each item? Explain why you selected the units you did.

# 2. Mass/Weight



Use the electronic balance/scale to mass the following materials. Place your measurements in the spaces below. Be sure to check the unit on the scale to verify that you are measuring in grams (g) not ounces (oz). Write the name of the unit above the columns below.

Mass of coin	g	mg	kg
Mass of dollar	g	mg	kg
Mass of empty 10 ml graduated cylinder	g	mg	kg
Mass of graduated cylinder with 10 ml water	g	mg	kg
Mass of 10 ml water	g	mg	kg

2a. What is the difference between weight and mass?

2b. Why are they used synonymously on Earth?

# 3. Volume (of liquids) $(1ml = 1cm^3)$

Use the glassware provided to measure the volume of the following containers. Place your measurements in the below. In the last column of the chart write which glassware you used to measure the liquid with. (10 ml graduated cylinder, 50 ml graduated cylinder, 50 ml beaker, beaker, 100 ml graduated cylinder)



Volume of the purple liquid	L	cl	ml	
Volume of water	L	cl	ml	
Total Volume of bottle	L	cl	ml	

3a. Do all of the glassware measure with the same degree of accuracy? Explain

3b. What determines which glassware you measure with?

3c. What is a meniscus? Why is it necessary to know about it when measuring liquids?

#### 4. Temperature

Use the thermometer to measure the temperature of the following items. Write the name of the unit measure above the column that contains its abbreviation. Circle the temperature/s which you were able to measure directly.



Ice water	°F	°C
Boiling water	°F	°C
Room temperature	°F	0°

4a. Is there a formula to convert between °F and °C?

4b. What is human body temperature?

#### Summary:

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2. The metric prefix denoting 1000x is \_\_\_\_\_\_.

3. If you are changing m to cm. what direction do you move the decimal point?

4. How many times larger is a centigram than a milligram?

5. What is the metric value for mass? \_\_\_\_\_

What is the metric value for length? \_\_\_\_\_\_

7. What is the prefix value for 100X? \_\_\_\_\_

8. What is the prefix value for 1/100? \_\_\_\_\_

9. If we are moving from a large value to a small value, we move the decimal point to the \_\_\_\_\_\_.

10. If we are moving the decimal point to the right we are moving from a \_\_\_\_\_\_ value to a

\_\_\_\_\_ value.

11. Would kg, ml, or cm signify a measurement of weight?

12. Would volume be measured with a meter stick, a graduated cylinder, or a balance scale?

13. What is the smallest unit of measurement shown on a meter stick? \_\_\_\_\_\_

14. Would you measure the height of the ceiling using m, mm, or cm<sup>3</sup>? \_\_\_\_\_

15.Convert 500 mL to liters.

16.Convert 5g to mg. \_\_\_\_\_

17. If an object is 100 mm in length, how many cm is it?

18.20 mL of pure water weighs how much? \_\_\_\_\_

19. If you were to fill a graduated cylinder to 250 mL, then drop an object in that raises the water level to 300 mL, what is the volume of the object, in cubic centimeters?